1 INTRODUCTION

New cermets based on NbC with alternative Ni binder phases and a Co/Ni combination, being face-centered cubic (FCC), emerged to minimize the consumption of traditional WC–Co alloys, mainly for the manufacture of cutting tools for machining subject to high temperatures and wear resistance.\(^1\) NbC, \(0.77 < y < 0.96\), has been used as a secondary hard phase for high-speed steels and as an inhibitor for the WC grain growth for cermet tools, but its potential to be used as a primary hard phase needs to be further explored.\(^3\) Little attention was given to NbC in the 19th century, but recent studies have shown promising results, allowing us to replace WC with NbC, especially for cutting tools and wear resistance.\(^{1,4}\) An NbC-based cermet with an amount of...
combined carbon being above 10 \%\textsubscript{w}, includes NbC + 12Ni + C (free), with its eutectic temperature close to 1265 °C, as seen on Figures 1a and 1b\textsuperscript{4}. The liquid phase of FCC (Ni) starts at this temperature for amounts of combined carbon of 9.33 \%\textsubscript{w} and 10 \%\textsubscript{w}. Below 9.25 \%\textsubscript{w}, there is slow cooling; temperatures of 1500 °C and 1100 °C result in equilibrium phases, NbC–liquid and NbC–NbNi\textsubscript{3}, respectively. With an amount of combined carbon of 9.25–9.75 \%\textsubscript{w}, when being cooled down from 1250 °C, the formation of an NbNi\textsubscript{3} compound occurs, and there is a pseudo-binary region NbC + FCC + NbNi\textsubscript{3}, with a very narrow range.\textsuperscript{3,4} NbC\textsubscript{y} can form sub-cubic carbides, allowing a wide range of stoichiometry of the combined carbon, from 6.5–11.8 \%\textsubscript{w}, and can be adapted by stoichiometry as shown in the Nb-C binary phase diagram, Figure 1d.\textsuperscript{5}

In terms of additive manufacturing, L–PBF is a promising alternative for refractory alloys where metal powders are selectively melted layer-by-layer using a computer-controlled laser beam.\textsuperscript{6} One advantage of the L–PBF process is the ability to manufacture components of complex shapes and small dimensions, something that cannot be easily manufactured with the conventional process of powder metallurgy. This technology has been used more frequently by several industrial sectors in order to reduce production steps, mainly for manufacturing mechanical components with complex geometries, and with a high value and low volume of production.\textsuperscript{7} The present work aims to compare the microstructure characteristics and properties of NbC–30Co, NbC–30Ni and NbC–30(Co, Ni) alloys, manufactured using the conventional liquid phase sintering (LPS) process and laser powder bed fusion (L–PBF), to point out the challenges in the additive manufacturing of the cermet based on NbC with alternative binding phases.

2 EXPERIMENTAL PART

This section presents the experimental procedures including the conventional (pressing and sintering) and selective laser melting AM (L–PBF) process. The following cemented carbide samples were prepared, with the compositions shown in Table 2: (I) NbC–30Co, (II) NbC–30Ni and (III) NbC–30(Co, Ni). High-purity com-

![Figure 1: a) Phase evolution with temperature of the studied NbC-12Ni cermet, b) Phase diagram simulation of the NbC-12Ni system, c) Isothermal section of Nb-C-Ni at 1420 °C, d) Binary Nb-C phase diagram](image)
Commercial Co, Ni and WC powders were used, with the following particle sizes: (2.0, 4.8 and 1.5) μm, respectively. The metal powders of NbC, Co, Ni and WC were mixed and homogenized by means of a conventional paddle stirrer for 2 h, using isopropyl alcohol to avoid breaking, crushing or reducing the particulates. Figure 2 shows the SE of each mixture (Table 2) of the constituent powders (Table 1) used in this work. The LPS samples were compacted, without polymeric lubricants, into a metallic matrix with a rectangular shape of (6 × 12 × 37) mm, using a pressure of 50–125 MPa. Green compacted samples were sintered in a vacuum oven (2 × 10–2 mbar), with sintering cycles at 1330 °C (1603 K) and 1370 °C (1643 K) and a heating rate of 10 °C min–1 (283 K/min), then stabilized for 60 min and slowly cooled inside the oven.

The L–PBF process was carried out in an Omni-Sint-160 machine, as shown in Figure 3a. In this machine, the powder bed is created by a vibrating device that has a rubber scraper ruler and a metallic roller. The rotational speed of the roller is independently controlled and should be zero. The area of the powder bed is too large (around 630 cm²), and the compaction force is applied by a 2 kg roller (19.6 N); thus, the compaction pressure is very low. This powder feeding device was developed specially for this L–PBF machine. The metallic

<table>
<thead>
<tr>
<th>Powder</th>
<th>Theoretical density (g·cm⁻³)</th>
<th>Average grain size (μm)</th>
<th>Purity (w%)</th>
<th>Apparent density (g·cm⁻³)</th>
<th>Origin</th>
</tr>
</thead>
<tbody>
<tr>
<td>NbC</td>
<td>7.750</td>
<td>1.0</td>
<td>99.650</td>
<td>3.0</td>
<td>F&amp;X Electro-Materials Limited (China)</td>
</tr>
<tr>
<td>WC</td>
<td>15.630</td>
<td>1.5</td>
<td>99.530</td>
<td>4.0</td>
<td>Buffalo Tungsten Inc. (USA)</td>
</tr>
<tr>
<td>Co</td>
<td>8.908</td>
<td>2.0</td>
<td>99.970</td>
<td>2.0</td>
<td>Nanjing Hanrui Cobalt Co. (China)</td>
</tr>
<tr>
<td>Ni</td>
<td>8.900</td>
<td>5.0</td>
<td>99.850</td>
<td>2.5</td>
<td>CVMR Corporation (Canada)</td>
</tr>
</tbody>
</table>

Table 2: Weight balance (w%) of the prepared alloys

<table>
<thead>
<tr>
<th>Alloy</th>
<th>Description of cermet</th>
<th>WC</th>
<th>NbC</th>
<th>Co</th>
<th>Ni</th>
<th>Theoretical density (g·cm⁻³)</th>
</tr>
</thead>
<tbody>
<tr>
<td>I</td>
<td>NbC–30Co</td>
<td>3.0</td>
<td>67.0</td>
<td>30.0</td>
<td>0.0</td>
<td>8.003</td>
</tr>
<tr>
<td>II</td>
<td>NbC–30Ni</td>
<td>3.0</td>
<td>67.0</td>
<td>0.0</td>
<td>30.0</td>
<td>8.005</td>
</tr>
<tr>
<td>III</td>
<td>NbC–30(Co, Ni)</td>
<td>3.0</td>
<td>67.0</td>
<td>15.0</td>
<td>15.0</td>
<td>8.004</td>
</tr>
</tbody>
</table>

Figure 2: SE images of the mixtures of metal powders: a) I NbC–30Co, b) II NbC–30Ni and c) III NbC–30(Co,Ni) alloy

Figure 3: a) OmniSint-160 machine, b) vibration device with a roller, c) sintering chamber, d) powder bed flowability & compressibility, e) sintering with different parameter levels P₁–ν₁
powder is fed manually into a container that has a capacity of 1500 g and is vibrating. It makes the powder go through a 35 mesh sieve. The translation and rotation movement simultaneously spreads and compacts the powder, generating a powder bed, Figure 3d. Thin layers are produced with the rollers and it is possible there is even extra fine metal powder, used for 3D printing additive manufacturing. During direct sintering, the chamber was inertized with a high-purity dynamic argon atmosphere (0.3 L min⁻¹), Figure 3c. The oxygen concentration during the process was around 300 μg/g. The deposition was limited to 40 layers, with a layer thickness of 30 μm. The shrinkage during consolidating was high, so the final sample thickness was smaller than 1200 μm.

Figure 4a shows a schematic of the process parameters for L-PBF, Figure 4b presents a 3D view for the L-PBF process, and Figure 4c presents the scanning strategy with a 67° rotation, from layer N to layer N-1. The laser power (P_L) ranged from 50–125 W, the scan speed (v_s) parameter ranged from 50–150 mm·s⁻¹, keeping the scan line spacing (h_s) equal to 80 μm. The layer

<table>
<thead>
<tr>
<th>T(°C)</th>
<th>1330 °C</th>
<th>1370 °C</th>
</tr>
</thead>
<tbody>
<tr>
<td>Alloy (I) Co Binder Phase</td>
<td>Type porosity A08 (0.6 φ%)</td>
<td>Type porosity A06 (0.2 φ%)</td>
</tr>
<tr>
<td>Type porosity B08 (0.6 φ%)</td>
<td>Phase γ – fine (irregular)</td>
<td>Phase γ – medium (irregular)</td>
</tr>
<tr>
<td>Alloy (II) Ni Binder Phase</td>
<td>Type porosity A08 (0.6 φ%)</td>
<td>Type porosity A04 (0.06 φ%)</td>
</tr>
<tr>
<td>Type porosity A06 (0.6 φ%)</td>
<td>Phase γ – fine (irregular)</td>
<td>Phase γ – coarse (regular)</td>
</tr>
<tr>
<td>Alloy (III) Co-Ni Binder Phase</td>
<td>Type porosity A08 (0.6 φ%)</td>
<td>Type porosity A02 (0.02 φ%)</td>
</tr>
<tr>
<td>Type porosity A06 (0.6 φ%)</td>
<td>Phase γ – fine (irregular)</td>
<td>Phase γ – coarse (regular)</td>
</tr>
</tbody>
</table>

Figure 5: Types of apparent porosities (ISO 4505) and microstructures (ISO 4499) of samples NbC–30Co, NbC–30Ni and NbC–30(Co, Ni) produced via the conventional LPS route (σ = 125 MPa)
3 RESULTS AND DISCUSSIONS

The apparent porosities (LM 100×) and microstructures of the samples obtained with light microscopy (LM) after the LPS process at temperatures of 1330 °C (1603 K) and 1370 °C (1643 K) are presented in Figure 5. During sintering, solid-solid bonding occurs between the NbC carbides, before the emergence of the liquid phase, forming necks between the particles and porosities. The Ni and Co–Ni binding phases of the samples sintered at 1370 °C indicate good wettability around the NbC grains. For the Co-based alloys, the NbC grains are not well rounded. In the NbC–Co alloy, density can be achieved in the solid-solid and solid-liquid state; however, the microstructures are not spheroidized and the mechanical properties are very low. Metallographic determination of the porosity and uncombined carbon was performed via LM (100×). The alloys manufactured via LPS at the temperature of 1330 °C show apparent porosities of types A08 and B08 (0.6–1.0%, 4000 pores cm<sup>-2</sup>) and at 1370 °C, they show porosities of types A02 to A06 (0.02–0.2%, 140 to 1300 pores cm<sup>-2</sup>), according to ASTM B276. The γ-phase is determined after surface etching with a Murakami reagent when the phase is light yellowish-brown and has a typically rounded shape. Its size is estimated and recorded as γ-fine, γ-medium or γ-coarse according to ISO 4499. The metallographic analysis shows a good granulometric and homogeneous distribution of NbC for the NbC–30Ni and NbC–30(Co, Ni) alloys. The microhardness values (HV1) are presented in Figure 5a for all 3 chemical compositions, (I) NbC–30Co alloy, (II) NbC–30Ni alloy and (III) NbC–30(Co, Ni) alloys.

Figure 5 presents the microstructures of the samples sintered at 1330 °C and 1370 °C. It is possible to observe, considering micrographs with a low magnification of 100 diameters (ISO 4505), that the level of porosity is particularly low for the NbC–30Ni and NbC–30(Co, Ni) samples. Micrographs with a low magnification of 1500 diameters (1 μm = 0.00003937 in) (ISO 4499) show that Ni is the best binder for NbC, providing a more homogeneous microstructure with a better distribution of the carbides. There are Co pools for NbC–30Co.

For the NbC–12Ni alloys, the eutectic temperature is 1265 °C. Figures 1a and 1b, while for the NbC–30Ni and NbC–30(Co, Ni) alloys, this temperature is higher, around 1350 °C. Thus, alloys with nickel or cobalt-nickel should be sintered at higher temperatures, higher than the sintering temperature used here (1370 °C) in order to obtain more homogeneous microstructures, with the carbide grains well dispersed in the binder phase. The hardness obtained for NbC–Co is lower than for WC–Co alloys, while higher values can be achieved with sintering temperatures higher than 1370 °C.

The sintering of conventional WC–Co hard metals with a high content of the binder phase is carried out in a temperature range of 1300–1380 °C. For this work, temperatures of 1330 °C and 1370 °C were chosen. Due to the low hardness of hard metals, depending on the binder phase (20–30 wt%), which varies from 600–900 HV, it is possible to carry out conventional machining of parts after sintering. The results are similar to those found in the literature for conventional cemented carbides WC–Co, showing the efficiency of obtaining the mixtures by means of weight balance. To obtain the Vickers microhardness (1 kgf), the values were calculated based on the ASTM C1327-15 standard, presenting the average of 10 microhardness values obtained with their respective standard deviations. Figures 6a to 6c show properties of the samples sintered with LPS, as a function of the temperatures of 1330 °C and 1370 °C.

The melting point of Ni equals 1455 °C, which is lower than that of cobalt at 1495 °C. However, for the sintered products NbC–(Co, Ni), obtained via LPS, to achieve a satisfactory densification, it is necessary to use a longer stabilization time and/or a higher temperature during the furnace LPS. The problem related to this is that nickel has a ten time greater vapor pressure than cobalt in the LPS process, which results in a considerable loss of the nickel binding phase and, therefore, it is necessary to control the working pressure. The loss of nickel in practice has been reported to range from 4–10 wt%, depending on the sintering temperature. It can be seen in Table 2 that the densities of the samples sintered at

![Figure 6](image-url)

**Figure 6:** a) Vickers hardness (HV1), b) Density (g·cm<sup>-3</sup>), c) Linear shrinkage (%) of NbC–30Co, NbC–30Ni and NbC–30(Co, Ni) samples as a function of compaction pressures (MPa) for the temperatures of 1330 °C and 1370 °C.
1370 °C exceeded the theoretical density, making it difficult to compare the relative densities of the NbC–30Ni and NbC–30(Co, Ni) samples. However, the NbC–30Ni and NbC–30(Co, Ni) alloys showed the best linear shrinkages, from 18–20\%/%, similar to those of the conventional WC–Co alloys, with high levels of the binder phase.

Among the samples processed via LPS, the NbC–30Ni alloy showed the mean microhardness values, HV 839 ± 67 (1330 °C) and HV 847 ± 21 (1370 °C), which were higher than those of alloys NbC–30Co, being HV 756 ± 150 (1330 °C) and HV 669 ± 100 (1370 °C) and NbC–30(Co, Ni), being HV 748 ± 56 (1330 °C) and HV 831 ± 30 (1370 °C), indicating that NbC–Ni hybrid particles are harder than NbC–Co particles and the alternative NbC–(Co,Ni) binding phase. Furthermore, when compared with traditional alloys, the microhardness of the NbC–30Ni alloy was higher than the hardness of WC–Co (HV 800), though this value may be overestimated. The hardness of conventional cemented carbides (hard metals) may vary from HV30 1000–2000 with the cobalt content varying from 10–20 w/\% for submicrograins (< 1 μm) and fine grains (> 2 μm), respectively. For NbC-based cermets with a high level of binder phase, 20 w/\%, the Vickers microhardnesses should vary from HV30 900–1100, depending on the average grain size of NbC and the addition of WC as the reinforcement to the metallic matrix. For WC-Co traditional alloys, the hardness ranges from HV800–2000 with the WC constituent ranging from 75–95 w/\% and the Co content ranging from 5–25 w/\%. These are produced with conventional molding methods, including injection molding, extrusion molding and powder metallurgy.

Investigations of the L–PBF process (direct sintering) included the creation of a \( P_L = v_s \) process function.
map, see Figure 4d, resulting in different hardness values for different energy levels (J·mm$^{-3}$). Figure 7. Figures 7a to 7c show the microindentation hardness behavior of the NbC–30Co, NbC–30Ni and NbC–30(Co,Ni) alloys as a function of the sintering strategy using PBF-L for different factor levels ($P_1$–$V_3$). For this reason, a load of 1 kgf was used for comparison with the samples sintered via the conventional route. Figure 6a. It can be seen that in Figure 7 the following colors were used in the tables: white, blue, green, yellow and red, correlating and differentiating the behavior of the microhardness of each sample of each group, based on previous experience, catalogs of manufacturers and researchers of traditional cemented carbide alloys. In Figure 7, the white color indicates an application for hot forming tools, with a microhardness lower than HV 600 (< 80 HRA), a binder content of 30–35 w/% and the average size of extra coarse grains (> 10 μm). The blue color indicates an application for rolls and cylinders for lamination, with a microhardness ranging from HV 600–800 (80–83 HRA), with a binder content of 25–30 w/% and the average grain size ranging from coarse to extra coarse (8–10 μm). The green color indicates an application for mechanical cold forming tools and die cutting tools for thin and medium sheets, with a microhardness ranging from HV 800–1000 (83–86 HRA), with a binder content of 18–23 w/% and the average grain size ranging from medium to coarse (5–7 μm). The yellow color indicates an application for wear resistant parts or components, with a microhardness ranging from HV 1100–1400 (87–90 HRA), with a binder content of 10–15 w/% and the average grain size ranging from fine to medium (2–4 μm). And finally, the red color indicates an application for inserts or machining inserts, with a microhardness ranging from HV 1500–2000 (90–94 HRA), with a binder content of 3–12 w/% and the average grain size ranging from sub-micron to fine (<1 μm). The relationship of the size and distribution of the apparent average grain size and metallographic determination of the microstructure in cemented carbides, for each application cited above, are based on ASTM B 390 and ISO 4499 standards. When selecting the compositions of refractory alloys based on NbC, or WC, and suitable processing parameters (LPS and L-PBF process) for hard metals, a wide combination of mechanical properties can be achieved. In particular, their combination of microhardness and toughness makes them attractive for many industrial applications. However, the variation in the binder content, carbide particle size and carbide particle distribution are critical factors that affect the mechanical properties of cermets, mainly the hardness values.

Figure 8 shows the porosity, cracks and microstructures (observed with the LM) of the NbC–30Co, NbC–30Ni and NbC–30(Co, Ni) samples produced with L–PBF. To visualize the porosities and cracks of the samples, the LM with a 100× magnification without a chemical etch was used. To reveal the microstructures, an electrochemical etch produced with a Murakami instrument was used at 3 V for 5 s. The hypotheses about the cracks, or embrittlement, of some samples may be related to the loss of the combined carbon content due to the presence of oxygen during sintering, detected by a gas sensor (300 μg/g), resulting in the formation of compounds, or brittle phases, and residual stresses during cooling. In addition, the evaporation of the binder phase occurs, which consequently decreases the size of the NbC grains and forms embrittlement compounds; for ex-
ample, for the NbC–30Ni alloy, there is a high probability for the formation of NbNi 3 and NbC compounds or phases. Figures 1a and 1d, due to the carbon loss and Ni evaporation, as well as an abnormal NbC grain growth. Figures 8a to 8c, seen in the microstructures of cermets obtained via L–PBF, especially for the Ni-based alloys above 330 J·mm −3. The formation of embrittlement phases in the NbC-based alloys, at specific points in the samples, is related to the loss of carbon, Figure 1d, due to the high energy involved in the L–PBF process. The presence of oxygen inside the sintering chamber contributes to the loss of carbon, which depends on the atmosphere during direct sintering and, therefore, the carbon mass balance cannot be fully controlled, decarburization is inevitable and, consequently, there are variations in the hardness for different energy levels. Moreover, uncertainties arise in the control of the microstructures, regarding the ternary phase diagrams of NbC alloys. The NbC–30 Ni and NbC–30(Co, Ni) alloy samples showed smaller amounts of microcracks when compared to the NbC–30Co alloy. However, they also showed the highest amount of macroporosity. The cracks may be related to the high thermal gradient and porosity with a combination of the power and scanning velocity of the laser beam. NbC particles are directly exposed to a laser beam, and due to the excess energy, they can melt and grow abnormally.

The NbC–30Co alloy samples showed higher hardness values compared to the NbC–30Ni and NbC–30(Co, Ni) samples, for all energy levels (J·mm −3). All samples number XIII (1000 J·mm −3) of the NbC–30Co, NbC–30Ni and NbC–30(Co, Ni) alloys, Figures 9a to 9c, respectively, were, in general, apparently dense, but microcracks and porosities of type A and B, from 10–25 μm, were observed. It was not possible to determine experimental densities because samples were very small (< 1 μm), and with irregular shapes and surfaces. We detected porosities and microcracks of the samples with different chemical compositions (NbC–30Co, NbC–30Ni and NbC–30(Co, Ni)), produced with the same volumetric energy density (1000 J·mm −3). Porosity was lower for NbC30Co, probably due to a better interaction of liquid Co with NbC particles. On the other hand, microcracks are related to high thermal gradients. The L–PBF samples had 40 layers; as l, was 30 μm, the expect thickness was 1200 μm. The average printed thickness was 802 ± 65 μm, so the shrinkage was 33.2 ± 6 %.

The microstructure obtained with the L–PBF process showed dendrites and abnormal growth of NbC for all samples, Figures 10a to 10c. In the LPS process, the redistribution of solute in the liquid and in the solid have the values predicted by the phase diagrams, but this only occurs with very slow cooling. However, using the L–PBF process, this does not occur under normal solidification conditions. These out-of-equilibrium conditions tend towards high microsegregation, emergence of dendrites Ni 3Nb and NbC abnormal grain growth. Figure 10d shows the solidification mode from the plane to the cell for the NbC-based carbide alloys. The effect of constitutional supercooling on the solidification mode makes it columnar dendritic and equiaxed dendritic. The greater the constitutional supercooling in the binding phase, the easier is the nucleation of columnar grains and equiaxed dendrites. According to the classical theory of eutectic growth, the growth, or solidification rate, and the thermal gradient at the solid/liquid interface are fundamental parameters for defining the solidification microstructure.

The obtained microstructures are evaluated with regard to their regularity and dimensions, and also in relation to the formation of primary NbC phases, NbC precipitations and the degeneration of the eutectic structure. The microstructural analysis of the samples processed via L–PBF with a varied thermal gradient and different growth rates revealed three distinct types of microstructure: eutectic-rich regions, columnar and equiaxed dendrites. We noticed the limitation of the magnification in the case of LM and the identification of microconstituents when compared with SEM, mainly when observing the interface of layer N–1 with layer N, Figures 10a and 10c. The existence of this coarse region in the contours, Figures 10b, indicates that there was solute segregation, necessary to adjust the volume fractions of the phases. The formation of a cellular microstructure or eutectic regions is associated with a combined effect of the impurities rejected from the solid phases, the growth rate and the thermal gradient imposed on the solid/liquid interface, resulting in constitutional supercooling ahead of the interface.

4 CONCLUSIONS

In summary, the main objective of this work was to compare the microstructures, porosities and hardness of the NbC–30Co, NbC–30Ni and NbC–30(Co,Ni) alloys obtained via LPS and L–PBF sintering processes. From the results obtained, the following conclusions are drawn:

• The solidification microstructures of cermet alloys obtained via L–PBF are very dependent on the temperature, energy and mainly on the cooling rate, with which the crystallographic orientation and morphology of NbC can be modified easily when compared to the LPS technique.
• All alloys obtained via L–PBF showed dendritic structures and an abnormal grain growth of NbC. At higher rates (Ev >400 J·mm −3), they transformed into heterogeneous microstructures, with regions rich in eutectic, columnar dendritic and equiaxed dendritic growth.
• The NbC–Ni and NbC–(Co,Ni) alloys obtained via LPS showed complete densification and a regular microstructure when compared with the NbC-Co al-
loy samples at 1370 °C kept in a vacuum oven for 60 min.

• The abnormal growth of NbC grains was due to dissolution and reprecipitation into larger NbC grains. The alloys containing the Co binder phase showed a higher hardness when compared to those containing Ni and Co-Ni.

• The indentation microhardness obtained via L–PBF increases due to a higher power and lower scanning speed of the laser and it is superior to the samples processed via LPS, with energies above 400 J·mm⁻³.

• Samples with high energy (Ev > 400 J·mm⁻³) exhibited irregular microstructures, high hardness and microcracks. Samples with low density energies (Ev < 300 J·mm⁻³) exhibited regular microstructures, but with microporosity, a lack of binder phase and a lack of fusion between the layers of deposition.

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5 REFERENCES


